Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

## Da-Tong Zhang, Fang-Gang Sun, Ya-Wei Sun, Gui-Yun Duan and Jian-Wu Wang*

School of Chemistry and Chemical Engineering, Shandong University, Jinan 250100, People's Republic of China

Correspondence e-mail: yugp2005@yahoo.com.cn

## Key indicators

Single-crystal X-ray study
$T=298 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.038$
$w R$ factor $=0.108$
Data-to-parameter ratio $=12.0$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
(C) 2006 International Union of Crystallography Printed in Great Britain - all rights reserved

## 1-(4,6-Dimethoxy-1,3,5-triazin-2-yloxy)-1 H-benzotriazole

In the title compound, $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{~N}_{6} \mathrm{O}_{3}$, the mean planes of the benzotriazole ring system and the 1,3,5-triazine ring make a dihedral angle of $89.3(3)^{\circ}$. The crystal packing is stabilized by $\pi-\pi$ stacking interactions and van der Waals forces.

## Comment

Carbodiimides combined with 1-hydroxy-1 $H$-benzotriazole (HOBt) have been widely employed in peptide synthesis (Chen et al., 1989). However, carbodiimides, which are necessary components in the formation of the activated ester, can cause allergic reaction (Bodanszky \& Williams, 1967). The title compound, (I) (Fig. 1), was synthesized to replace the combination of carbodiimides and HOBt. In this paper, we report its crystal structure.


The bond lengths and angles of the benzotriazole and 1,3,5triazine systems (Table 1) are in agreement with the values reported earlier (Xu et al., 2005; Główka \& Iwanicka, 1989). The mean planes of the benzotriazole ring system and the 1,3,5-triazine ring (C9-C11/N4-N6) make a dihedral angle of 89.3 (3) ${ }^{\circ}$. The crystal packing of (I) (Fig. 2) is stabilized by van der Waals forces and $\pi-\pi$ stacking interactions between the 1,3,5-triazine rings [the $C g \cdots C g^{i}$ distance is 3.573 (6) $\AA$, where $C g$ is the centroid of the C9-C11/N4-N6 ring] and between the benzotriazole ring systems of neighbouring molecules [the distance between the centroids of the C7/C8/N1-N3 and C3$\mathrm{C} 8^{\mathrm{ii}}$ rings is 3.810 (2) $\AA$ [symmetry codes: (i) $-x,-y, 1-z$; (ii) $\left.\frac{1}{2}-x, \frac{1}{2}-y, 1-z\right]$.

## Experimental

The title compound was synthesized by the reaction of 1-hydroxy- 1 H benzotriazole ( 0.01 mol ) and 2-chloro-4,6-dimethoxy-1,3,5-triazine ( 0.01 mol ) in the presence of $N$-methylmorpholine ( 20 ml ) at room temperature ( 5 h ). Purification was achieved by recrystallization from

Received 9 December 2005 Accepted 3 January 2006 Online 7 January 2006


Figure 1
View of (I), with displacement ellipsoids at the $40 \%$ probability level.


A packing diagram for (I).
a mixture of hexane/dichloromethane ( $1: 1 \mathrm{v} / \mathrm{v}$ ) in $92 \%$ isolated yield $(2.52 \mathrm{~g})$. Crystals suitable for X-ray diffraction analysis were obtained by slow evaporation of a solution in a mixture of hexane/dichloromethane $(1: 1 \mathrm{v} / \mathrm{v})$ at room temperature for one week.

## Crystal data

| $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{~N}_{6} \mathrm{O}_{3}$ | $D_{x}=1.480 \mathrm{Mg} \mathrm{m}^{-3}$ |
| :--- | :--- |
| $M_{r}=274.25$ | Mo $K \alpha$ radiation |
| Monoclinic, $C 2 / c$ | Cell parameters from 4441 |
| $a=21.602(3) \AA$ | reflections |
| $b=7.3765(12) \AA$ | $\theta=2.2-25.8^{\circ}$ |
| $c=18.248(3) \AA$ | $\mu=0.11 \mathrm{~mm}^{-1}$ |
| $\beta=122.139(2)$ | $T=298(2) \mathrm{K}$ |
| $V=2462.1(7) \AA^{\circ}$ | Block, colourless |
| $Z=8$ | $0.28 \times 0.25 \times 0.12 \mathrm{~mm}$ |

$D_{x}=1.480 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 4441
reflections
$\theta=2.2-25.8^{\circ}$
$\mu=0.11 \mathrm{~mm}^{-1}$
$T=298(2) \mathrm{K}$
Block, colourless
$0.28 \times 0.25 \times 0.12 \mathrm{~mm}$

## Data collection

| Bruker SMART CCD area-detector | 2166 independent reflections |
| :--- | :--- |
| $\quad$ diffractometer | 1772 reflections with $I>2 \sigma(I)$ |
| $\varphi$ and $\omega$ scans | $R_{\text {int }}=0.026$ |
| Absorption correction: multi-scan | $\theta_{\max }=25.0^{\circ}$ |
| $\quad(S A D A B S ;$ Sheldrick, 1996 $)$ | $h=-25 \rightarrow 23$ |
| $\quad T_{\min }=0.969, T_{\max }=0.989$ | $k=-8 \rightarrow 8$ |
| 6143 measured reflections | $l=-21 \rightarrow 21$ |

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.039$
$w R\left(F^{2}\right)=0.109$
$S=1.06$
2166 reflections
181 parameters
H -atom parameters constrained

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0659 P)^{2}\right. \\
& \quad+0.215 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.17 \mathrm{e}^{-3} \AA^{-3} \\
& \Delta \rho_{\min }=-0.14 \mathrm{e}^{-3}
\end{aligned}
$$

Table 1
Selected geometric parameters ( $\left(\AA^{\circ}{ }^{\circ}\right.$ ).

| O3-N2 | $1.3751(15)$ | N4-C10 | $1.298(2)$ |
| :--- | :--- | :--- | :--- |
| O3-C10 | $1.3753(17)$ | N4-C9 | $1.3418(19)$ |
| N1-N3 | $1.2960(18)$ | N5-C10 | $1.3174(19)$ |
| N1-N2 | $1.3394(18)$ | N5-C11 | $1.3201(18)$ |
| N2-C8 | $1.3495(19)$ | N6-C9 | $1.3164(19)$ |
| N3-C7 | $1.378(2)$ | N6-C11 | $1.3352(19)$ |
|  |  |  |  |
| N2-O3-C10 | $114.63(11)$ | N3-N1-N2 | $106.90(12)$ |

All H atoms were placed in calculated positions, with $\mathrm{C}-\mathrm{H}=0.93$ and $0.96 \AA$, and included in the final cycles of refinement using a riding model, with $U_{\mathrm{iso}}(\mathrm{H})=1.2 U_{\mathrm{eq}}(\mathrm{C})$ for the aryl H atoms and $1.5 U_{\mathrm{eq}}(\mathrm{C})$ for the methyl H atoms.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

## References

Bodanszky, M. \& Williams, N. J. (1967). J. Am. Chem. Soc. 89, 685-689. Bruker (1998). SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
Bruker (1999). SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
Chen, S. T., Wu, S. H. \& Wang, K. T. (1989). Synthesis, pp. 37-38.
Główka, M. L. \& Iwanicka, I. (1989). Acta Cryst. C45, 1765-1767.
Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Xu, L. Z.,Yang, S. H., Zhu, C. Y., Li, K. \& Liu, F. Q. (2005). Acta Cryst. E61, o259-o260.

